This Page Is Inserted by IFW Operations and is not a part of the Official Record

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

IMAGES ARE BEST AVAILABLE COPY.

As rescanning documents will not correct images, please do not report the images to the Image Problem Mailbox.

UK Patent Application GB GB GB 2 104 087 A

- (21) Application No 8222933
- (22) Date of filing 9 Aug 1982
- (30) Priority data
- (31) 291649
- (32) 10 Aug 1981
- (33) United States of America (US)
- (43) Application published 2 Mar 1983
- (51) INT CL³ C08G 18/10 18/66//C08J 5/00
- (52) Domestic classification
 C3R 32A 32D16B 32D16C
 32D6C 32D6J 32D6K
 32D6L 32E2A 32E2Y 32E6
 32F5 32G2Y 32J1A
 32J1X 32J1Y 32J2C
 32J2D 32J2F 32J2X
 32J2Y 32J3A 32J3B
 32J3Y 32KH 32KL 32S
 C11 C12 C25 L2X L5D L6D
 PD PX SP
 U18 1458 1570 1660
 3054 C3R
- (56) Documents cited GB 1521686 GB 1520417 GB 1351773
- (58) Field of search
- (71) Applicants
 The University of
 Liverpool,
 (Great Britain),
 P.O. Box 147,
 Liverpool,
 L69 3BX
 Ethicon Inc.,
 (USA—New Jersey),
 US Route No. 22,
 Somerville,
 New Jersey 08876,
 United States of America
 - (72) Inventors

 Denis Keith Gilding,

 Roy Malcolm Clarke
 - (74) Agents
 Carpmaels and Ransford,
 43 Bloomsbury Square,
 London,
 WC1A 2RA

(64) Proc ss for the manufacture of p lyurethane resin f r lectrostatic spinning

(57) A process for the preparation of polyurethane resins particularly suitable for electrostatic spinning which comprises forming a prepolymer of polyether glycol and a disocyanate reacted in a molar ratio of about 1:2, and thereafter reacting the prepolymer with additional

diis cyanate and a diol extender, the final millar ratio of the polyether, isocyanate and dill reactants being approximately 1:3.7:2.5. The reaction is conducted in an organic solvent at a temperature of about 45°C and is continued with incremental additions of organic solvent until a polymer having an inherent viscosity of from about 2.0 to 2.4 in a solution of 25 to 30 percent solids is obtained. The solution is diluted to about 15 percent solids prior to spinning.

GB 2 104 087 A



SPECIFICATION
Process for the manufacture of polyurethane resin for electrostatic spinning

field fthe inv nti n

This inventin relates to a process for preparing polyurethane resins, and more particularly to a process for preparing polyurethane resins which are especially suitable for spinning into fibers by the electrostatic
 spinning process.

Background of the invention

Electrostatic spinning of solutions of fiberforming polymers is well known. Electrostatic spinning of polyurethane solutions to form fine diameter fibers is also known as described for example in U.S. Patent 4,043,331 and U.S. Patent 4,044,404.

The process of electrostatic spinning of polyurethanes and other fiber-forming polymers 20 involves the introduction of a solution of the polymer into an electrostatic field where droplets of the solution are attenuated into fibers while being drawn to an electrode where the fibers are collected. The spinning environment is controlled 25 to evaporate the solvent while the fibers are being formed in the electrostatic field so that the fibers collected on the electrode retain their individual identity. Some fusing between fibers at points of intersection may occur, and is often desirable, as a result of residual solvent content in the fibers. Eventually, residual solvent is removed from the mat of collected fibers as part of the total spinning process.

The fibers obtained by the electrostatic
spinning process are thin, generally in the order of
0.1 to 25 microns, and preferably 0.5 to 10
microns. Fibers having different characteristics
are obtained by adjusting the spinning conditions
including the molecular weight and concentration
of polymer in the spinning solution. In all cases
however, it is essential that the spinning solution
be a composition which is capable of being
attenuated in the electrostatic field. Some
problems have been experienced in producing
fibers of uniform diameter from certain polymer
solutions, including solutions of polyurethanes.

It is accordingly an object of the present invention to provide a polyurethane polymer which is particularly suitable for use in the electrostatic spinning of fine fibers. It is a further object of this invention to provide a process for the preparation of the aforesaid polyurethane polymers. These and other objects of the present invention will be apparent from the ensuing description and claims.

Summary of the Invention

Polyurethane polymers are prepared by solution polymerization in a two stage process wherein polytetramethylene ether glycol is first reacted with diphenylmethane diisocyanate (MDI) in molar proportions of about 1:2 and in an rganic solvent t form a prep lymer. The

prepolymer is subsequently extended by reaction with butane diol and additional MDI in a molar proportions of about 2.5 and 1.6 (based on glycol) respectively. Polymerization is continued with incremental additions of organic solvent until a 25 to 30% by weight solution of polymer having an inherent visc sity of from about 2.0 t 2.4 is 70 obtain d, at which p int the reaction is terminated by the additi n f a small amount of methan I.

The resulting polyurethane polymer solution is diluted to about 15% solids with a volatile organic solvent and readily spun into fine fibers of a uniform diameter by the electrostatic spinning process. The resulting porous fibrous mat in the form of a sheet or tube has many desirable properties including uniformity of fibers and pire size.

Description of preferred embodiments

The process of the present invention is illustrated by the following example:

150 g (0.075 m) polytetramethylene ether glycol (PMEG) (molecular weight 2000) is charged molten to a reaction vessel equipped with a heater, stirrer and thermometer. The molten PMEG is stirred vigorously under 150 mm Hg vacuum at 100—150°C for 1-1/2 hours to remove moisture and then allowed to cool under a dry nitrogen atmosphere to 80°C.

39 g (0.156 m) of flake diphenyl methane disocyanate (MDI) is added quickly to the reaction vessel followed by 60 g freshly distilled dimethyl formamide (DMF). The mixture is stirred continuously at 45°C±2° for approximately 2 hours under an atmosphere of dry nitrogen to form a prepolymer.

30.61 g (0.122 m) MDI, 16.88 g (0.1875 m) of dry butane diol (BD) and 180 g DMF are quickly added to the prepolymer with stirring, taking precautions to exclude moisture. The reaction is continued under a dry nitrogen atmosphere at 45°C±2° with continuous stirring.

As the viscosity of the solution increases to 400—500 poise, 115 g of additional DMF are added. This procedure is repeated three times t give a final polymer solution of 28.8% solids. After the final addition of DMF, the reaction is continued until a viscosity of 600 poise at 45°C is attained, at which time the reaction is terminated by the addition of 2 ml methanol. The inherent viscosity of the polymer is determined to be 2.16 in HFIP at 0.1 g/dl concentration. The complete process takes approximately 12 hours.

To prepare the polymer for electrostatic spinning, the 28.8% solids polymer solution is diluted with methyl ethyl ketone to obtain a final spinning solution of approximately 15 percent solids by weight.

In the process of the present invention, the preferred molar proportion of PMEG to MDI in the first stage reaction for the preparation of the prepolymer is 1:2. In accordance with the present invention, it is essential that the molar excess of MDI in this reaction not exceed 5 percent, and the

preferred limits are from about 90% up to 105% with an optimum range of from about 96% to 100%. In the preceding example, the molar proportion of MDI was 2.08, providing a 4% excess f the preferred molar amount. By thus -ilmiting the amount of free MDI during the preparation of the prepolymer, and by carrying out the reaction at the relatively low temperature of 45°C, it is believed that the possibility of cross-10 linking in the prepolymer is substantially reduced,

and that this results in the improved electrostatic spinning performance observed with the polymers of the present invention.

The final molar proportions of the 15 PMEG:MDI:BD reactants used in the process of the present invention are preferably about 1:3.7:2.5. Minor variations in these proportions in the order of plus or minus 5 percent are permissible without departing from the essence of 20 the present invention.

The polyether glycol used in the present invention is preferably PMEG having a molecular weight in the range of 1500-2500, and most preferably about 2000, although higher and lower molecular weight compositions can be used. 25 Other polyether glycols known to be useful in the preparation of fiber-forming polyurethanes which may also be used in the present invention include polyethylene ether glycol, polypropylene ether glycol, 1,2-polydimethylene ether glycol, and

polydecamethylene ether glycol. The diisocyanates useful in the present invention are also those known to be useful in the preparation of fiber-forming polyurethanes. The most common compound is MDI as used in the preceding example, but other organic diisocyanates may also be used with good results, including the aromatic, aliphatic and cycloaliphatic diisocyanates. Representative 40 compounds include 2,4-tolylene diisocyanate, mphenylene diisocyanate, 4-chloro-1,3-phenylene diisocyanate, 4,4'-biphenylene diisocyanate, 1,5naphthylene diisocyanate, 1,4-tetramethylene diisocyanate, 1,6-hexamethylene diisocyanate,

45 1,10-decamethylene diisocyanate, 1,4cyclohexylene diisocyanate, 4,4'-methylene-bis-(cyclohexyl isocyanate) and 1,5tetrahydronaphthylene diisocyanate. Arylene diisocyanates, i.e., those in which each of the two 50 isocyanate groups is attached directly to atomatic ring, are preferred.

The diol extenders used in the present invention in the molar proportions as set forth above are also the common diol extenders used 55 in the polyurethane art. Although the most common compound is 1,4-butane diol as used in the preceding example, any of the other known diol extenders may also be used with good results.

The first stage reaction of the polyether glycol 60 and diisocyanate is conducted in solution of about 75% solids by weight. The second stage reaction for the preparation of the polyurethane resin is initiated with additional solvent to reduce the solids level to about 50% by weight. As the

second stage reacti in proceeds, increases in viscosity are monitored by the relative decrease in stirrer speed or increase in power requirements, and additional solvent is added in increments 70 maintain the viscosity at less than about 400-500 poise until the silds level is decreased to 25 to 30 percent by weight. The reaction is then allowed to proceed with visc sity monit red by sampling the reaction vessel until the polymer attains an inherent viscosity of from about 2.0 t 2.4, and most preferably of about 2.2.

The organic solvent used in the polymerization process is preferably DMF although other v latile organic solvents such as dimethylacetamide 80 (DMAc) or methyl ethyl ketone (MEK) are also suitable. The use of MEK as a diluent in the final spinning solution is desirable to increase the volatility of the solvent and assure its rapid evaporation during formation of the fibers.

85 Spinning conditions, particularly temperature and air flow, may have to be increased to promote evaporation of the solvent where higher boiling solvents such as DMAc are used. For details n electrostatic spinning procedures, reference is made to U.S. 4,043,331, particularly Example 5 thereof which deals with spinning a solution of a commercial polyurethane polymer.

Claims

1. A process for the preparation of 95 polyurethane resins comprising:

a. reacting in an organic solvent a polyether glycol and a diisocyanate in a molar ratio of about 1:2 to form a prepolymer.

b. reacting said prepolymer with additional 100 diisocyanate and a diol extender in an organic solvent to form a polyurethane resin, the final molar ratios of the polyether glycol, diisocyanate and diol being about 1:3.7:2.5,

c. continuing said reaction with incremental 105 additions of organic solvent until a 25 to 30 percent by weight solids solution of said polyurethane resin having an inherent viscosity of from about 2.0 to 2.4 is obtained, and

d. terminating said reaction by the addition of 110 an alcohol.

2. A process of claim 1, wherein said reactions are conducted at 45°C±2°.

3. A process of claim 1 or claim 2, wherein said polyether glycol is polytetramethylene ether 115 glycol.

4. A process of any one of claims 1 to 3, wherein said dilsocyanate is diphenyl methane diisocyanate.

5. A process of any preceding claim, wherein 120 said diol extender is 1,4-butane diol.

6. A process of any preceding claim, wherein said organic solvent is dimethylformamide.

7. The process of any preceding claim, wherein said diisocyanate is present in an amount of from about 90% to not more than 105% of said molar ratio of 1:2 during formation of the prepolymer.

8. The process of any preceding claim, wherein said reaction of polyether glycol and diisocyanate

is conducted in a solution of about 75 wt percent solids.

- 9. The process of any preceding claim, wherein said reaction of said prepolymer is initially in a silution of about 50 wt percent solids.
- 10. The process of claim 9, wherein the incremental addition of solvent during the continuation of the reaction is effective to maintain the viscosity of the solution below about 500 polse until the solids level is reduced to 25 to 30 percent by weight and the reaction is allowed to go to the termination point.
 - 11. A process for the preparation of polyurethane resins substantially as hereinbefore described with reference to the foregoing example.
 - 12. A polyurethane resin produced by the process of any preceding claim.

- 13. A process for the preparation of
 polyurethane resin solutions for electrostatic
 spinning comprising the process of any one of
 claims 1 to 11 and the further step of diluting the
 polyurethane resin solution t about 15 percent
 by weight solids with a second organic solvent.
- 14. The process of claim 13, wherein said second organic solvent is dimethylformamide.
 - 15. The process of claim 13, wherein said second organic solvent is methyl ethyl ketone.
- 16. A process for the preparation of 30 polyurethane resin solutions for electrostatic spinning substantially as hereinbefore described with reference to the foregoing example.
- 17. A polyurethane resin solution for electrostatic spinning produced by the process of any one of claims 13 to 16..